

(19)



Europäisches Patentamt
European Patent Office
Office européen des brevets



(11)

EP 1 078 896 A1

(12)

EUROPEAN PATENT APPLICATION

(43) Date of publication:
28.02.2001 Bulletin 2001/09

(51) Int Cl.7: **C03C 10/00, G02B 5/28**

(21) Application number: 00108432.6

(22) Date of filing: 18.04.2000

(84) Designated Contracting States:
**AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU
MC NL PT SE**
Designated Extension States:
AL LT LV MK RO SI

- Kataoka, Mariko
Sagamihara-shi, Kanagawa-ken (JP)
- Polensky, Donald G.
Morgan Hill, California 95037 (US)

(30) Priority: 10.08.1999 JP 22694799

(71) Applicant: **KABUSHIKI KAISHA OHARA**
Kanagawa-ken (JP)

(74) Representative:
Jönsson, Hans-Peter, Dr.Dipl.-Chem. et al
Patentanwälte
von Kreisler Selting Werner,
Bahnhofsvorplatz 1 (Deichmannhaus am Dom)
50667 Köln (DE)

(72) Inventors:
• Goto, Naoyuki
Machida-shi, Tokyo-to (JP)

Remarks:

Amended claims in accordance with Rule 86 (2)
EPC.

(54) **Glass-ceramics for a light filter and a light filter**

(57) Glass-ceramics for a light filter capable of preventing variation of refractive index in a band-pass filter have a coefficient of thermal expansion within a range from $95 \times 10^{-7}/^{\circ}\text{C}$ to $140 \times 10^{-7}/^{\circ}\text{C}$ within a temperature range from -20°C to $+70^{\circ}\text{C}$ and, preferably, Young's modulus of 85GPa and bending strength of 10kg/mm²,

and light transmittance for plate thickness of 10mm of 60% or over within a wavelength range from 950nm to 1600nm. Predominant crystal phases of these glass-ceramics should preferably be (a) lithium disilicate and (b) one selected from the group consisting of α -quartz, α -quartz solid solution, α -cristobalite and α -cristobalite solid solution.

EP 1 078 896 A1

Description

[0001] This invention relates to glass-ceramics for a light filter and also to a light filter using glass-ceramics

[0002] There are light filters which cut or pass light of a specific wavelength and there are also light filters which reduce intensity of light without depending upon wavelength. The former includes a band-pass filter which passes only a specific wavelength, a notch pass filter which cuts a specific wavelength and high-pass and low-pass filters which pass only wavelengths shorter or longer than a specific wavelength. The latter includes an ND filter.

[0003] Light filters can be classified also into an absorption type filter and an interference type filter. A representative absorption type filter is the ND filter and a representative interference type filter is the band-pass filter. A substrate made of plastic is used for absorption type filters such as those for photography. Since a substrate for light filters which are subject to a strong laser beam requires durability and heat resistance property, amorphous glass is exclusively employed for such substrate.

[0004] The band-pass filters are made by forming, on a substrate made of, e.g., glass, a multi-layer film of dielectric by alternately laminating an H layer of a dielectric thin film having a high refractive index and an L layer of a dielectric thin film having a low refractive index.

[0005] In a band-pass filter which is used for the WDM(wavelength division multiplexing) optical communication system, temperature stability of the center wavelength of the band poses a problem when a narrow band width for passing wavelengths is set for applying the band-pass filter to a wavelength of a higher density. More specifically, the band-pass filter is a sensitive element in which the center frequency of the band varies even with a slight variation in temperature and, therefore, temperature compensation should be made by a temperature controller when the band-pass filter is used. Such temperature controller, however, cannot actually be employed because of limitation in the space where the band-pass filter is located. The temperature stability has become a matter of increasing importance since it is necessary to reduce the band width as the amount of light information increases.

[0006] In the past, amorphous glass has been used as a substrate for the band-pass filter. This prior art substrate is not sufficient in its compressive stress to the film and its durability since its thermal expansion property and mechanical strength are not sufficiently high. Further, amorphous glass has a low surface hardness and, moreover, a relatively large amount of alkali ingredient must be added if a high thermal expansion property is to be provided and this poses a problem of elution of alkali ingredient during and after forming of the dielectric film on the substrate. Thus, amorphous glass cannot sufficiently satisfy the demands for a substrate for a light filter, particularly a substrate for a band-pass filter.

[0007] It is, therefore, an object of the invention to provide a material suitable for a substrate for a light filter which has eliminated the above described disadvantages of the prior art substrate and has a thermal expansion property which is sufficient for avoiding variation in the refractive index at a temperature at which a filter formed with a mono-layer or multi-layer film is used (i.e., having a high coefficient of thermal expansion and thereby imparting compressive stress to the film to improve temperature stability of the refractive index of the film) and also has a mechanical property which imparts sufficient durability to the filter and further has excellent light transmittance.

[0008] It is, another object of the invention to provide a light filter made of such substrate.

[0009] Accumulated studies and experiments made by the inventors of the present invention for achieving the above described objects of the invention have resulted in the finding, which has led to the present invention, that glass-ceramics having a coefficient of thermal expansion, mechanical strength and light transmittance within specific ranges are suitable for achieving these objects of the invention.

[0010] According to the invention, there are provided glass-ceramics for a light filter having a coefficient of thermal expansion within a range from $95 \times 10^{-7}/^{\circ}\text{C}$ to $140 \times 10^{-7}/^{\circ}\text{C}$ within a temperature range from -20°C to $+70^{\circ}\text{C}$.

[0011] In one aspect of the invention, the glass-ceramics have Young's modulus of 85GPa or over.

[0012] In another aspect of the invention, the glass-ceramics have bending strength of 10kg/mm² or over.

[0013] In another aspect of the invention, light transmittance for plate thickness of 10mm is 60% or over within a wavelength range from 950nm to 1600nm.

[0014] In another aspect of the invention, the glass-ceramics have a composition which comprises, in weight percent:

SiO ₂	70 - 77%
Li ₂ O	8 - 12%
K ₂ O	0.5 - 3%
MgO + ZnO + SrO + BaO	1 - 5%
P ₂ O ₅	1.5 - 3%
ZrO ₂	2 - 7%
Al ₂ O ₃	3 - 9%
Sb ₂ O ₃ + As ₂ O ₃	0 - 2%.

[0015] In another aspect of the invention, the glass-ceramics contain, as predominant crystal phases,

(a) lithium disilicate, and

(b) at least one selected from the group consisting of α -quartz, α -quartz solid solution, α -cristobalite and α -cristobalite solid solution.

[0016] In another aspect of the invention, the glass-ceramics are substantially free of Na_2O and PbO .

[0017] In another aspect of the invention, there is provided a light filter which is made by forming a dielectric film on the above described glass-ceramics.

[0018] In still another aspect of the invention, there is provided a light filter which is made by forming a dielectric film on glass-ceramics having a larger coefficient of thermal expansion than dielectric which constitutes the dielectric film.

[0019] Reasons for limiting the thermal expansion property, Young's modulus, bending strength, light transmittance, composition and predominant crystal phases of the glass-ceramics for a light filter according to the invention will be described below. The composition of the glass-ceramics is expressed on the oxide basis (in weight percent) as in their base glasses.

[0020] Description will be made first about thermal expansion property. As described previously, the temperature stability of the center wavelength of the band is very important and glass-ceramics having a larger coefficient of thermal expansion than a material which constitutes the film are required. The studies and experiments conducted by the inventors of the present invention have resulted in the finding that, if the coefficient of thermal expansion within the temperature range from -20°C to $+70^\circ\text{C}$ is $95 \times 10^{-7}/^\circ\text{C}$ or over, sufficient compression stress can be imparted to the film within a temperature range in which the glass-ceramics are used as a band-pass filter and that, if the coefficient of thermal expansion exceeds $140 \times 10^{-7}/^\circ\text{C}$, difference in the coefficient of thermal expansion between the substrate and the filter becomes so large that problems such as separation of the film from the substrate take place. A preferable range of the coefficient of thermal expansion is $110 \times 10^{-7}/^\circ\text{C}$ to $130 \times 10^{-7}/^\circ\text{C}$ and a more preferable range thereof is $120 \pm 5 \times 10^{-7}/^\circ\text{C}$.

[0021] In the band-pass filter, the temperature stability of the center wavelength depends to some extent on refractive index temperature coefficient of a dielectric which constitutes the thin film and, to a larger extent than that, on a coefficient of thermal expansion of the substrate. This is because refractive index is determined by a film atomic density of the thin film. That is, the higher the film atomic density of the thin film is, the smaller becomes variation caused by the temperature of the center frequency. The film atomic density of the thin film is greatly influenced by the coefficient of thermal expansion of the substrate for the light filter on which the thin film is formed. More specifically, the temperature of the substrate during the film forming process becomes about 200°C and the substrate thereby is considerably expanded. The thin film is formed on this expanded substrate and, as the substrate is cooled, the thin film is subjected to compressive stress due to difference in the coefficient of thermal expansion between them. As a result, the film atomic density of the thin film increases and the refractive index thereby increases. As the coefficient of thermal expansion of the substrate increases, the compressive stress applied to the dielectric thin film formed on the substrate increases with the result that variation in the refractive index due to temperature at which the filter is used decreases. For this reason, it is desirable to set the coefficient of thermal expansion of the glass-ceramics at a larger value than the coefficient of thermal expansion of the dielectric thin film.

[0022] Having regard to severe conditions in which the band-pass filter is used, strength against mechanical deformation, i.e., bending strength and Young's modulus of the glass-ceramics, in addition to the above described properties, cannot be ignored.

[0023] As regards Young's modulus, since the substrate is processed to a small chip ($2\text{mm} \times 2\text{mm} \times 2\text{mm}$ or below) after forming of a thin film, a high Young's modulus and a high bending strength are required for the substrate. Having regard to such subsequent processing, the glass-ceramic should preferably have Young's modulus of 85GPa or over and bending strength of 10kg/mm^2 or over.

[0024] As regards light transmittance, if light transmittance is low, inconveniences such as reduction in the signal-to-noise ratio will take place in producing signals. It is therefore desirable that light transmittance should be as large as possible and it has been found that light transmittance of 60% at the minimum is necessary. The wavelength range used for the band-pass filter is 950nm to 1600nm and light transmittance of 60% or over for plate thickness of 10mm is required for this wavelength range. As to light transmittance within this wavelength range, light transmittance should preferably be 75% or over and, more preferably, 80% or over.

[0025] As regards the crystal phases grown in the glass-ceramics, the glass-ceramics should preferably contain, as their predominant crystal phases, (a) lithium disilicate, and (b) at least one selected from the group consisting of α -quartz, α -quartz solid solution, α -cristobalite and α -cristobalite solid solution. By causing these crystal phases to grow in the glass-ceramics, the bending strength and Young's modulus can be markedly increased and the coefficient of thermal expansion within the temperature range from -20°C to $+70^\circ\text{C}$ can be brought to the range from $95 \times 10^{-7}/^\circ\text{C}$ to $140 \times 10^{-7}/^\circ\text{C}$. Further, light transmittance at a plate thickness of 10mm of 60% or over can be realized within the

wavelength range from 950nm to 1600nm in which the band-pass filter is used.

[0026] Reasons for limiting the composition range of the base glass as described above will now be described.

[0027] The SiO_2 ingredient is a very important ingredient in that, by heating a base glass, it forms lithium disilicate, α -quartz, α -quartz solid solution, α -cristobalite and α -cristobalite solid solution which constitute the predominant crystal phases of the glass-ceramics. If the amount of this ingredient is below 70%, the crystal phases which have grown in the glass-ceramics are instable and their texture tends to become too rough whereas if the amount of this ingredient exceeds 77%, difficulty arises in melting and forming the base glass.

[0028] The Li_2O ingredient is a very important ingredient in that, by heating the base glass, it forms lithium disilicate as one of the predominant crystal phases. If the amount of this ingredient is below 8%, difficulty arises in growing of this crystal phases and also in melting of the base glass whereas if the amount of this ingredient exceeds 12%, the crystal phase grown is instable and its texture tends to become too rough and, moreover, chemical durability deteriorates.

[0029] The K_2O ingredient is effective for improving the melting property of the glass and preventing the texture of the crystal phase grown from becoming too rough. A preferable amount of this ingredient is 0.5% or over. An excessive amount of this ingredient, however, causes the texture of the crystal grown to become too rough, causes change in the crystal phase and deteriorates chemical durability and, for this reason, the amount of this ingredient should preferably be restricted to 3% or below.

[0030] The MgO , ZnO , SrO and BaO ingredients are effective for improving the melting property of the glass, preventing the texture of the crystal grown from becoming too rough, and enabling adjustment of light transmittance by adjusting refractive index of the glass phase which constitutes the matrix phase. If the total amount of these ingredients is below 1%, these effects cannot be achieved whereas if the total amount of these ingredients exceeds 5%, the crystal phases obtained are instable and the texture thereof tends to become too rough.

[0031] The P_2O_5 ingredient is indispensable as a nucleating agent. For achieving this effect, a preferable amount of this ingredient is 1.5% or over. For preventing devitrification of the base glass and maintaining stability in the large scale production, the amount of this ingredient should be limited to 3% or below.

[0032] The ZrO_2 ingredient is a very important ingredient which, like P_2O_5 ingredient, functions as a nucleating agent for the glass and also is effective for making the grown crystal grains finer and improving mechanical strength and chemical durability of the material. For achieving these effects, a preferable amount of this ingredient is 2% or over. If an excessive amount of this ingredient is added, difficulty arises in melting the base glass and a material such as ZrSiO_4 is left unmelted. For this reason, the amount of this ingredient should be limited to 9% or below.

[0033] The Al_2O_3 ingredient is an ingredient which improves chemical durability and mechanical strength, particularly bending strength, of the glass-ceramics. For achieving the objects of the invention, it is necessary to add 3% or over, preferably 4% or over, of this ingredient. If an excessive amount of this ingredient is added, the melting property and resistance to devitrification of the glass deteriorate and, moreover, β -spodumene and β -cristobalite grow as crystal phases. The β -spodumene and β -cristobalite are crystals having a very small coefficient of thermal expansion and growth of these crystals significantly reduces a coefficient of thermal expansion of the glass-ceramics obtained. Therefore, a preferable amount of this ingredient is 9% or below and, preferably, 8% or below.

[0034] The Sb_2O_3 and As_2O_3 ingredients may be added as a refining agent. Addition of a total sum of one or both of these ingredients up to 2% will suffice. A preferable total sum is 1% or less.

[0035] The reason for substantially excluding Na_2O and PbO in the glass-ceramics will be described. In forming a multi-layer film on a substrate, Na_2O contained in a material poses a problem because Na ions are eluted into the multi-layer film resulting in deterioration of the film properties. As to PbO , this ingredient is undesirable from the standpoint of protecting the environment and use of this ingredient should be avoided to the maximum extent possible.

[0036] For manufacturing the glass-ceramics for a light filter according to the invention, the base glass having the above described composition is melted, is subjected to heat forming and/or cold forming, is heat treated for producing a crystal nucleus under a temperature within a range from 500°C to 600°C for about one to seven hours, and further is heat treated for crystallization under a temperature within a range from 700°C to 780°C for about one to seven hours.

[0037] The glass-ceramics thus obtained by the heat treatment had, as their predominant crystal phases, (a) lithium disilicate, and (b) at least one selected from the group consisting of α -quartz, α -quartz solid solution, α -cristobalite and α -cristobalite solid solution.

[0038] By lapping and polishing the glass-ceramics thus obtained by a conventional method, glass-ceramic substrates for a light filter having surface roughness (Ra) (arithmetic mean roughness) within a range from 1.0Å to 5.0 Å are provided. The glass-ceramics according to the invention are suitable for an interference type light filter in which a multi-layer film of dielectric is formed on a glass-ceramic substrate, particularly, for a band-pass filter having a dielectric multi-layer film formed by alternately laminating a dielectric thin film (H layer) having a high refractive index and a dielectric thin film (L layer) having a low refractive index.

[0039] As the dielectric, inorganic oxides such as TiO_2 , Ta_2O_5 , Nb_2O_5 and SiO_2 may preferably be used. In a band-pass filter used for the wavelength range from 950nm to 1600nm, combinations of $\text{TiO}_2/\text{SiO}_2$, $\text{Ta}_2\text{O}_5/\text{SiO}_2$ and $\text{Nb}_2\text{O}_5/\text{SiO}_2$

SiO₂ may preferably be used as combinations of the H layer and the L layer.

[0040] The light filter of the present invention can be provided by forming a dielectric thin film on the surface of the glass-ceramic substrates. For forming the thin film, deposition, RF ion plating, magnetron sputtering, plasma ion plating etc. may be employed. Among them, deposition is particularly preferable.

[0041] Examples of the invention will now be described. Tables 1, 2 and 3 show Examples No. 1 to No. 8 of the glass-ceramics for a light filter according to the invention and a comparative example of a prior art glass substrate for a light filter with respect to their composition, grown crystal phases, coefficient of thermal expansion, Young's modulus, bending strength and light transmittance. In the tables, lithium disilicate is abbreviated as "LD".

Table 1

	Examples		
	1	2	3
SiO ₂	76.3	73.9	77.2 (Wt.%)
Li ₂ O	9.9	9.0	9.5
P ₂ O ₅	2.0	2.1	2.0
ZrO ₂	2.3	4.0	1.3
Al ₂ O ₃	7.0	6.0	4.0
MgO	0.8	0.5	0.8
ZnO	0.5	0.5	0.5
SrO		2.5	
BaO			2.5
K ₂ O	1.0	1.3	2.0
Sb ₂ O ₃	0.2	0.2	0.2
As ₂ O ₃			
Predominant crystal phase	LD α-quartz	LD α-quartz	LD α-quartz α-cristobalite
Coefficient of thermal expansion (×10 ⁻⁷ /°C) (-20°C to +70°C)	110	122	130
Young's modulus(GPa)	98	100	90
Bending strength(kg/mm ²)	11	20	15
Light transmittance(%) (950 - 1600nm)	75	80	65

Table 2

	Examples		
	4	5	6
SiO ₂	76.6	73.9	77.2 (Wt.%)
Li ₂ O	9.9	9.0	9.0
P ₂ O ₅	2.0	2.1	2.0
ZrO ₂	2.3	4.0	1.3
Al ₂ O ₃	7.0	6.0	4.0
MgO	0.5	0.5	0.8
ZnO	0.5	0.5	0.5
SrO		1.5	0.5
BaO		1.0	2.5
K ₂ O	1.0	1.3	2.0
Sb ₂ O ₃	0.2	0.2	0.2
As ₂ O ₃			
Predominant crystal phase	LD α-quartz	LD α-quartz	LD α-quartz

Table 2 (continued)

	Examples		
	4	5	6
			α -cristobalite
Coefficient of thermal expansion ($\times 10^{-7}/^{\circ}\text{C}$) (-20°C to $+70^{\circ}\text{C}$)	119	127	130
Young's modulus(GPa)	98	100	90
Bending strength(kg/mm ²)	15	23	12
Light transmittance(%) (950 - 1600nm)	85	80	62

Table 3

	Examples		Comparative Example 1
	7	8	
SiO ₂	73.9	75.3	45.87(Wt.%)
Li ₂ O	8.5	9.0	NaO=5.0
P ₂ O ₅	2.1	2.2	B ₂ O ₃ =4.7
ZrO ₂	4.2	4.5	TiO ₂ =3.88
Al ₂ O ₃	6.0	5.5	3.7
MgO	0.5	0.5	
ZnO	0.5	0.5	
SrO	2.5	1.0	
BaO	1.0		24.45
K ₂ O	0.6	1.3	12.3
Sb ₂ O ₃	0.2	0.2	0.1
As ₂ O ₃			
Predominant crystal phase	LD α -quartz	LD α -quartz	
Coefficient of thermal expansion ($\times 10^{-7}/^{\circ}\text{C}$) (-20°C to $+70^{\circ}\text{C}$)	120	122	93
Young's modulus(GPa)	94	99	75
Bending strength(kg/mm ²)	28	25	5
Light transmittance(%) (950 - 1600nm)	85	83	90

[0042] For manufacturing the glass-ceramics of the above described Examples No. 1 to No. 8, materials including oxides, carbonates and nitrates were weighed and mixed and molten in a conventional melting apparatus at a temperature within the range from about 1350°C to 1450°C. The molten glass was stirred to homogenize it and thereafter formed into a predetermined shape and annealed to provide a formed glass. Then, the formed glass was subjected to heat treatment to produce the crystal nucleus under a temperature within the range from 500°C to 600°C for about one to seven hours and then was subjected to a further heat treatment for crystallization under a temperature within the range from 700°C to 780°C for about one to seven hours to produce the desired glass-ceramics. Then, these glass-ceramics were lapped with diamond pellets of 800 - 2000# for 5 minutes to 30 minutes and were finally polished with cerium oxide polishing grains having an average diameter of 0.02 μm to 3 μm for 30 minutes to 60 minutes. Surface roughness (Ra) (roughness at the center line) was 5Å or below.

[0043] Crystal phases grown were identified by an X-ray Diffractometer (XRD) and an Energy Dispersive X-ray Spectroscopy (EDS).

[0044] Comparing Examples No. 1 to No. 8 with Comparative Example 1, the coefficient of thermal expansion of the

prior art glass substrate for a light filter was $93 \times 10^{-7}/^{\circ}\text{C}$ which was not sufficient for imparting compressive stress to a formed film. The prior art glass substrate exhibited also low Young's modulus of 75GPa and bending strength of 5kg/mm². In contrast, the glass-ceramics of the present invention had a sufficient coefficient of thermal expansion for imparting compressive stress to the formed film and also sufficient Young's modulus and bending strength which shows

that the glass-ceramics of the present invention are suitable for substrates for a light filter.

[0045] Interference type light filters which were provided by forming multi-layer films of $\text{TiO}_2/\text{SiO}_2$, $\text{Ta}_2\text{O}_5/\text{SiO}_2$ and $\text{Nb}_2\text{O}_5/\text{SiO}_2$ on the glass-ceramic substrates of the above described Examples had an excellent temperature stability of its center wavelength and were found to be most suitable for a band-pass filter for optical communication.

[0046] As described in the foregoing, according to the invention, the disadvantages of the prior art substrates are overcome and glass-ceramic substrates for a light filter having an excellent temperature stability of a center wavelength are provided. These features, i.e., high light transmittance, high thermal expansion property, high Young's modulus and high bending strength, are suitable for an interference type filter, particularly a band-pass filter and are most suitable for WDM and DWDM (density wavelength division multiplexing) in optical communication systems. Further, the band-pass filter elements which are provided by forming multi-layer dielectric films of $\text{TiO}_2/\text{SiO}_2$, $\text{Ta}_2\text{O}_5/\text{SiO}_2$ and $\text{Nb}_2\text{O}_5/\text{SiO}_2$ on the glass-ceramic substrates of the invention have an excellent temperature stability of the center wavelength and can be used not only for optical communication systems on the ground but also for space-based satellites.

Claims

1. Glass-ceramics for a light filter having a coefficient of thermal expansion within a range from $95 \times 10^{-7}/^{\circ}\text{C}$ to $140 \times 10^{-7}/^{\circ}\text{C}$ within a temperature range from -20°C to $+70^{\circ}\text{C}$.
2. Glass-ceramics as defined in claim 1 which have Young's modulus of 85GPa or over.
3. Glass-ceramics as defined in claim 1 or 2 which have bending strength of 10kg/mm² or over.
4. Glass-ceramics as defined in any of claims 1 to 3 wherein light transmittance for plate thickness of 10mm is 60% or over within a wavelength range from 950nm to 1600nm.
5. Glass-ceramics as defined in any of claims 1 to 4 wherein the glass-ceramics have a composition which comprises, in weight percent:

SiO_2	70 - 77%
Li_2O	8 - 12%
K_2O	0.5 - 3%
$\text{MgO} + \text{ZnO} + \text{SrO} + \text{BaO}$	1 - 5%
P_2O_5	1.5 - 3%
ZrO_2	2 - 7%
Al_2O_3	3 - 9%
$\text{Sb}_2\text{O}_3 + \text{As}_2\text{O}_3$	0 - 2%.

6. Glass-ceramics as defined in any of claims 1 to 5 which contain, as predominant crystal phases,
 - (a) lithium disilicate, and
 - (b) at least one selected from the group consisting of α -quartz, α -quartz solid solution, α -cristobalite and α -cristobalite solid solution.
7. Glass-ceramics as defined in any of claims 1 to 6 which are substantially free of Na_2O and PbO .
8. A light filter which is made by forming a dielectric film on glass-ceramics as defined in any of claims 1 to 7.
9. A light filter which is made by forming a dielectric film on glass-ceramics having a larger coefficient of thermal expansion than dielectric which constitutes the dielectric film.

Amended claims in accordance with Rule 86(2) EPC.

1. Glass-ceramics for a light filter having a coefficient of thermal expansion within a range from $95 \times 10^{-7}/^{\circ}\text{C}$ to $140 \times 10^{-7}/^{\circ}\text{C}$ within a temperature range from -20°C to $+70^{\circ}\text{C}$, having light transmittance for plate thickness of 10mm which is 60% or over within a wavelength range from 950nm to 1600nm, and containing, as predominant crystal phases,

(a) lithium disilicate, and

(b) at least one selected from the group consisting of α -quartz, α -quartz solid solution, α -cristobalite and α -cristobalite solid solution.

2. Glass-ceramics as defined in claim 1 which have Young's modulus of 85GPa or over.

3. Glass-ceramics as defined in claim 1 or 2 which have bending strength of 10kg/mm² or over.

4. Glass-ceramics as defined in any of claims 1 to 3 which are substantially free of Na₂O and PbO.

5. Glass-ceramics as defined in any of claims 1 to 4 wherein the glass-ceramics have a composition which comprises, in weight percent:

SiO ₂	70 - 77%
Li ₂ O	8 - 12%
K ₂ O	0.5 - 3%
MgO + ZnO + SrO + BaO	1 - 5%
P ₂ O ₅	1.5 - 3%
ZrO ₂	2 - 7%
Al ₂ O ₃	3 - 9%
Sb ₂ O ₃ + As ₂ O ₃	0 - 2%

6. A light filter which is made by forming a dielectric film on glass-ceramics as defined in any of claims 1 to 5.

7. A light filter which is made by forming a dielectric film on glass-ceramics having a larger coefficient of thermal expansion than dielectric which constitutes the dielectric film[.], said glass-ceramics having a coefficient of thermal expansion within a range from $95 \times 10^{-7}/^{\circ}\text{C}$ to $140 \times 10^{-7}/^{\circ}\text{C}$ within a temperature range from -20°C to $+70^{\circ}\text{C}$ and having Young's modulus of 85GPa or over.



European Patent
Office

EUROPEAN SEARCH REPORT

Application Number
EP 00 10 8432

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.7)
X	EP 0 924 171 A (OHARA KK) 23 June 1999 (1999-06-23) * page 15, line 11 - line 12 * TIH-53 OHARA-made glass * table V * * page 17, line 15 * ---	1-4,8	C03C10/00 G02B5/28
X	EP 0 781 731 A (YAMAMURA GLASS CO LTD) 2 July 1997 (1997-07-02) Comparative example 2 ---	1-7	
X	US 5 872 069 A (ABE MASAHIRO) 16 February 1999 (1999-02-16) * examples * Comparative examples * column 5, line 49 - line 60 * * column 8, line 44 * * claim 6 * * column 9, line 54 - line 62 * * column 12, line 11 * * claim 6 * ---	1-7	
X	GB 2 284 655 A (GEC ALSTHOM LTD) 14 June 1995 (1995-06-14) * examples 2,3; tables I,II * ---	1-7	C03C G02B
X	US 5 483 628 A (BORRELLI NICHOLAS F ET AL) 9 January 1996 (1996-01-09) * column 5, line 21 - line 25 * * column 8, line 4 - line 5 * * table II * ---	1	
X	US 4 940 636 A (BROCK LIESELOTTE ET AL) 10 July 1990 (1990-07-10) * column 2, line 15 - line 31 * * column 7, line 54 - column 8, line 3 * ---	9	
		-/--	
The present search report has been drawn up for all claims			
Place of search MUNICH		Date of completion of the search 15 June 2000	Examiner Somann, K
<p>CATEGORY OF CITED DOCUMENTS</p> <p>X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document</p> <p>T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document</p>			

EPF FORM 1500 03/02 (1/94/CO1)



European Patent
Office

EUROPEAN SEARCH REPORT

Application Number
EP 00 10 8432

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.7)
A	<p>EP 0 442 801 A (NIPPON TELEGRAPH & TELEPHONE) 21 August 1991 (1991-08-21)</p> <p>* page 5, line 5 - line 10 *</p> <p>* page 5, line 16 - line 18 *</p> <p>* page 5, line 32 - line 36 *</p> <p>* page 7, line 4 - line 9 *</p> <p>* page 7, line 51 - line 56 *</p> <p>* page 8, line 23 - line 24 *</p> <p>* page 8, line 57 - page 9, line 4 *</p> <p>* claim 2 *</p>	9	
			TECHNICAL FIELDS SEARCHED (Int.Cl.7)
The present search report has been drawn up for all claims			
Place of search		Date of completion of the search	Examiner
MUNICH		15 June 2000	Somann, K
<p>CATEGORY OF CITED DOCUMENTS</p> <p>X : particularly relevant if taken alone</p> <p>Y : particularly relevant if combined with another document of the same category</p> <p>A : technological background</p> <p>O : non-written disclosure</p> <p>P : intermediate document</p> <p>T : theory or principle underlying the invention</p> <p>E : earlier patent document, but published on, or after the filing date</p> <p>D : document cited in the application</p> <p>L : document cited for other reasons</p> <p>& : member of the same patent family, corresponding document</p>			

EPO FORM 1502 03.92 (P4/C01)

**ANNEX TO THE EUROPEAN SEARCH REPORT
ON EUROPEAN PATENT APPLICATION NO.**

EP 00 10 8432

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on
The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

15-06-2000

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP 0924171 A	23-06-1999	JP 11240736 A	07-09-1999
EP 0781731 A	02-07-1997	DE 69601697 D	15-04-1999
		DE 69601697 T	22-07-1999
		JP 10226532 A	25-08-1998
		US 5691256 A	25-11-1997
		US 5874376 A	23-02-1999
US 5872069 A	16-02-1999	JP 10188260 A	21-07-1998
GB 2284655 A	14-06-1995	NONE	
US 5483628 A	09-01-1996	AU 688149 B	05-03-1998
		AU 3783895 A	30-05-1996
		CA 2161494 A	26-05-1996
		CN 1133268 A	16-10-1996
		DE 69504461 D	08-10-1998
		DE 69504461 T	14-01-1999
		EP 0716050 A	12-06-1996
		JP 8239235 A	17-09-1996
		US 5537505 A	16-07-1996
US 4940636 A	10-07-1990	CN 1031605 A, B	08-03-1989
		DE 3853970 D	20-07-1995
		EP 0300579 A	25-01-1989
		ES 2077562 T	01-12-1995
		HU 50395 A	29-01-1990
		JP 1105203 A	21-04-1989
		JP 2740653 B	15-04-1998
		KR 9703196 B	15-03-1997
EP 0442801 A	21-08-1991	DE 69109513 D	14-06-1995
		DE 69109513 T	18-01-1996
		JP 2608633 B	07-05-1997
		JP 4211203 A	03-08-1992
		US 5332535 A	26-07-1994
		US 5234772 A	10-08-1993

EPO FORM P0459

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82